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are also beginning to replace it. **chemistry unit 8 - review - stark science** - modeling chemistry 1 u8 quiz 3
v2.0 chemistry unit 8 - review for the following questions 1. write the balanced equation and set up the bca
table. 2. determine the values that go into the table. do mole calculations on the side. 3. determine the
quantities you are asked to find. 1. sodium metal combines with chlorine gas to form sodium chloride.
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of all chemical reactions, including the role of energy. explain how a balanced equation represents these
features (include an example). **chemistry - unit 8 review - ==mr montero chemistry ...** - chemistry - unit
8 review molar volume and gas stoichiometry 1. a mixture contains 8.00 g each of o2, co2, and so2 at stp.
calculate the volume of this mixture. which of the gases would exert the greatest pressure and why? molar
volume = 22.4l 1mole 2. nitrogen gas and hydrogen gas combine to produce ammonia gas (nh3). what **unit 8
ws 1 mole relationships answers** - 1. mole -> mass conversions - using molar mass of each substance,
convert the following quantities. a. 10.0 mol cr 520 g f. 0.160 mol h2o 2.88 g. b. **modeling chemistry
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stoichiometry i - learning goal: students can check your answers (1) 4.0 g h2. (2) 228 g na. stoichiometry
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volume of a mole of gas purpose the purpose of this experiment to determine the volume of a mole of gas at
standard temperature and pressure. procedure the pictures below will help you follow the discussion of the
procedure. fig 1 strip of mg ribbon, stopper and copper wire. fig 2 the piece of mg wrapped in a cage of copper
wire. **name date pd chemistry unit 8 more practice in writing and ...** - modeling chemistry 1 u8 ws bal
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chemical equations for the following reactions. 1. zinc reacts with hydrogen chloride to form zinc chloride. 2.
sodium oxide reacts with water to form sodium hydroxide. 3. **myhotlunchprogram modeling chemistry u8
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